

Year:10 Subject: Chemistry (Trilogy)

**IMPLEMENTATION**

<p><b>INTENT</b> (including key concepts and skills)</p>	<p><b>Half Term 1</b> <b>Context:</b> C3 Chemical Changes</p> <p><b>Key Vocabulary:</b> Metal ore, reactivity series, oxidation, reduction, displacement reaction, pH scale, neutralisation, indicator</p> <p><b>Prior Learning</b> Y7 R1 Acids and Alkalis Y8 R4 Reactivity Y9 R5 Reactions with Metals</p> <p><b>Cultural Capital:</b> How commonly used metals are extracted</p>	<p><b>Half Term 2</b> <b>Context:</b> C3 Electrolysis</p> <p><b>Key Vocabulary:</b> Electrolysis, electrolyte, electrode, anode, cathode, aqueous solutions, half equation, oxidation and reduction in terms of electrons, aqueous solution, aluminium, molten</p> <p><b>Prior Learning</b> C2 Ionic Compounds</p> <p><b>Cultural Capital:</b> How useful aluminium is extracted and the environmental implications</p>	<p><b>Half Term 3</b> <b>Context:</b> C3 Energy Changes</p> <p><b>Key Vocabulary:</b> Energy transfer, exothermic, endothermic, reaction profile, bond energy</p> <p><b>Prior Learning</b> C2 Covalent Bonding</p> <p><b>Cultural Capital:</b> Everyday applications of exothermic and endothermic reactions</p>	<p><b>Half Term 4</b> <b>Context:</b> C4 Rate of reaction</p> <p><b>Key Vocabulary:</b> Rate of reaction, gradient, tangent, particle, collision theory, activation energy surface area, concentration, pressure, temperature, catalyst</p> <p><b>Prior learning:</b> Y7 P1 Particles and States of Matter C2 Reaction profiles and activation energy</p> <p><b>Cultural Capital:</b> How conditions can change the rate of everyday reactions</p>	<p><b>Half Term 5</b> <b>Context:</b> C4 Rate of reaction</p> <p><b>Key Vocabulary:</b> Rate of reaction, gradient, tangent, particle, collision theory, activation energy surface area, concentration, pressure, temperature, catalyst</p> <p><b>Prior learning:</b> Y7 P1 Particles and States of Matter C2 Reaction profiles and activation energy</p> <p><b>Cultural Capital:</b> How catalysts can be used to improve the rate and profit of chemical reactions.</p>	<p><b>Half Term 6</b> <b>Context:</b> C4 Reversible Reactions and Equilibrium</p> <p><b>Key Vocabulary:</b> Reversible reaction, hydrated, anhydrous, equilibrium, Le Chatelier's Principle (HT only)</p> <p><b>Prior Learning</b> C3 Endothermic and Exothermic Reactions</p> <p><b>Cultural Capital:</b> Use of anhydrous copper sulfate as a test for water. How altering conditions in industrial processes can affect yield (HT only)</p>
<p>Applying knowledge and understanding to explain observations</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>
<p>Use different types of scientific enquiry to answer scientific questions</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>
<p>Use technical terminology with confidence accuracy and precisely</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>
<p>Apply mathematical knowledge to scientific understanding</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>
<p>Aware of some of the social and economic implications of science</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>	<p>X</p>
<p><b>IMPACT</b></p>	<p>Assessment: At end of C3</p>	<p>Assessment: At end of C3</p>	<p>Assessment: C3 End of Topic Test</p>	<p>Assessment: At end of C4</p>	<p>Assessment: At end of C4</p>	<p>Assessment: C4 End of Topic Test Summer Summative on Paper 1 Content.</p>

Cultural Capital is the body of knowledge a student needs so that they can flourish in the future and not be left behind.

